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(54) Title: OXIDATION-STABLE PROTEASES (57) Abstract Novel <i>Bacillus</i> proteases with improved stability performance in solutions containing hypochlorite or other oxidizing agents.		

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OXIDATION-STABLE PROTEASES

FIELD OF INVENTION

This invention is in the field of proteases derived from strains of Bacillus sp. More specifically, the invention is directed towards a novel protease derived from a strain of a novel Bacillus sp., which is characterized by being stable in solutions containing hypochlorite and/or other oxidizing agents. Moreover, the invention is directed towards a process for the preparation of the protease, and the use of the protease in processes in which water containing hypochlorite is being used.

BACKGROUND OF THE INVENTION

Proteases have been marketed for more than 20 years for a lot of different purposes, the most important as being ingredients in detergents.

Proteases have been developed by isolation of proteases found in nature. Most commercially available proteases are obtained from the genus Bacillus. Currently new types of proteases enter the market, offering the possibility of giving a better cost/performance ratio at various specified conditions.

Examples of commercial Bacillus protease products are Alcalase®, Esperase®, Primase®, Savinase® and Durazyme® (a protein-engineered variant of Savinase), all available from Novo Nordisk A/S, Denmark. These and similar enzyme products from other commercial sources are active in detergent solutions, i.e. at pH values in the range from 8 to 11 and in the presence of sequestering agents, surfactants and bleaching agents such as sodium borate, but their activity decreases if the process water used contains hypochlorite. This is an increasing problem as more and more water in the industrialized world gets chlorinated.

SUMMARY OF THE INVENTION

It is an object of the present invention to provide novel proteases with improved stability performance in solutions containing hypochlorite.

5 Accordingly, in its first aspect, the invention provides a protease having immunochemical properties identical to those of a protease derived from the strain Bacillus sp., DSM 8473, the protease being stable in solutions containing hypochlorite.

10 In a second aspect, the invention relates to a biologically pure culture of a strain of a novel Bacillus sp. In a more specific aspect, the invention relates to a strain of Bacillus sp., DSM 8473, or a mutant or a variant thereof.

In a third aspect, the invention provides a process
15 for the preparation of the protease, which process comprises cultivation of a protease producing strain of a novel Bacillus sp. in a suitable nutrient medium, containing carbon and nitrogen sources and inorganic salts, followed by recovery of the desired enzyme. In a more specific aspect, Bacillus sp.,
20 DSM 8473, or a mutant or a variant thereof encoding a protease having immunochemical properties identical to those of the protease derived from Bacillus sp., DSM 8473, is cultivated.

In a fourth aspect, the use of the protease in processes in which water containing hypochlorite is being used,
25 is claimed.

BRIEF DESCRIPTION OF DRAWINGS

The present invention is further illustrated by reference to the accompanying drawings, in which

Fig. 1 shows the relation between temperature and the
30 proteolytic activity of a novel protease according to the invention (the protease preparation obtained according to Ex.1, with 2% of casein as substrate and at pH 9.5).

Fig.2 shows the relation between pH and the proteolytic activity of a novel protease according to the invention

(the protease preparation obtained according to Ex.1, with 2% of casein as substrate and at 25°C, using Britten-Robinson buffers adjusted to predetermined pH values in the pH range of from 6 to 11).

5 DETAILED DISCLOSURE OF THE INVENTION

The Microorganism

The novel microorganism of the invention, able to produce an enzyme of the invention, is represented by the strain that was isolated from a soil sample.

10 The novel Bacillus sp. has been deposited according to the Budapest Treaty on the International Recognition of the Deposits of Microorganisms for the Purpose of Patent Procedures, on 23 August 1993, at Deutsche Sammlung von Mikroorganismen und Zellkulturen GmbH, under Accession No. DSM 8473.

15 The microorganism of this invention is an aerobic, spore forming bacterium belonging to the genus Bacillus. Morphologically it can be described as motile rods with a diameter of 0.6-0.8 μm , and a length of 1-3 μm . The spores are cylindrical to ellipsoid, not swelling the sporangium, central
20 to subterminal. Optimum temperature for growth is within 30-50°C, and optimal pH for growth is within 6-8, good growth at 50°C. The microorganism forms yellow colonies, only when grown at 37°C - otherwise colourless, slimy colonies on nutrient agar slants, and no diffusion of pigment into the agar is observed.

25 Cultivation of the Microorganism

The microorganism of the invention can be cultivated under aerobic conditions in a nutrient medium containing assimilable carbon and nitrogen together with other essential nutrients, the medium being composed in accordance with the
30 principles of the known art.

Suitable carbon sources are carbohydrates such as sucrose, glucose and starch, or carbohydrate containing materials such as cereal grain, malt, rice and sorghum. The carbohydrate concentration incorporated in the medium may vary

widely, e.g. up to 25% and down to 1-5%, but usually 8-10% will be suitable, the percentages being calculated as equivalents of glucose.

The nitrogen source in the nutrient medium may be of
5 inorganic and/or organic nature. Suitable inorganic nitrogen sources are nitrates and ammonium salts. Among the organic
nitrogen sources quite a number are used regularly in fermentation processes involving the cultivation of bacteria. Illustrative examples are soybean meal, cotton seed meal, peanut meal,
10 casein, corn, corn steep liquor, yeast extract, urea and albumin. In addition, the nutrient medium should also contain usual trace substances.

For cultivation in tank fermentors it is necessary to use artificial aeration. The rate of aeration is similar to
15 that used in conventional tank fermentation.

After fermentation, liquid enzyme concentrates may be produced by removal of coarse material from the broth or, if desired, concentration of the broth by evaporation at low temperature or by reverse osmosis. Finally, preservatives may
20 be added to the concentrate.

Solid enzyme preparations may be prepared from the purified and/or concentrated broth by precipitation with salts, such as Na_2SO_4 or water-miscible solvents, such as ethanol or acetone. Removal of the water in the broth by suitable drying
25 methods, such as spray-drying, may also be employed.

Assay for Proteolytic Activity

The proteolytic activity is determined with casein as substrate. One Casein Protease Unit (CPU) is defined as the amount of enzyme liberating 1 mM of primary amino groups
30 (determined by comparison with a serine standard) per minute under standard conditions, i.e. incubation for 30 minutes at 25°C and pH 9.5.

The Enzymes

The enzymes of the invention are novel proteases.
35 They are alkaline proteases, obtainable by cultivation of a

microorganism of the invention, preferably Bacillus sp., DSM 8473, or a mutant or a variant thereof, in a suitable nutrient medium, containing carbon and nitrogen sources and inorganic salts. The enzymes can also be obtained by recombinant DNA-
5 technology.

The proteases of the invention can be described by the following characteristics.

Physical-chemical Properties

A molecular weight of 30 kD, determined by SDS-PAGE.
10 A pI of about 8.8 as determined by isoelectric focusing on LKB Ampholine® PAG plates.

The protease activity is inhibited by PMSF and Turkey-egg-white proteinase inhibitor. EDTA and soybean-protein inhibitor do not influence the protease activity.

15 The temperature activity relationship was determined with 2% casein as substrate and at pH 9.5. The assay for proteolytic activity described previously was used with the modification that the incubation temperature was varied in the interval of from 15 to 70°C. The result for a novel protease
20 is shown in Fig. 1. It appears from the figure that the protease possesses proteolytic activity at temperatures of from 15°C to 70°C, and have a temperature optimum within the range of from 50° to 60°C, around 60°C.

The dependence of activity on pH was determined by
25 the same procedure, using buffers adjusted to predetermined pH values in the pH range of from 6 to 11. The result is shown in Fig. 2. It appears from this figure that the enzyme possesses proteolytic activity at all pH values in this range (below 11 to above 6).

30 The proteases of the invention possess especial potentials in water containing hypochlorite. Ex. 2 illustrates this very clearly. In general the proteases have a residual activity of at least 45%, preferably above 60%, most preferably above 80% at 5 ppm NaOCl, and a residual activity of at least
35 10%, preferably above 20% at 10 ppm NaOCl.

Immunochemical Properties

The immunochemical properties can be determined immunologically by cross-reaction identity tests. The identity tests can be performed by the well-known Ouchterlony double
5 immunodiffusion procedure or by tandem crossed immunoelectrophoresis according to I. M. Roitt; Immunology, Gower Medical Publishing (1985) and N. H. Axelsen; Handbook of Immunoprecipitation-in-Gel Techniques; Blackwell Scientific Publications (1983), Chapters 5 and 14. The terms "antigenic identity" and
10 "partial antigenic identity" are described in the same book, Chapters 5, 19 and 20.

Monospecific antiserum was generated according to the above mentioned method by immunizing rabbits with one of the purified proteases of the invention. The immunogen was mixed
15 with Freund's adjuvant and injected subcutaneously into rabbits every second week. Antiserum was obtained after a total immunization period of 8 weeks, and immunoglobulin was prepared therefrom as described by N. H. Axelsen, supra.

Ouchterlony double immunodiffusion tests showed
20 immunochemical non-identity between the protease of the invention and the known alkaline serine proteases Savinase, Esperase, Durazyme, Primase (available from Novo Nordisk A/S), and Kazusase™ (available from SHOWA DENKO). A partial immunochemical identity was demonstrated with Alcalase from
25 Bacillus licheniformis.

Oxidizing agents

The proteases of the invention are stable against oxidizing agents such as hypochlorite, hydrogen peroxide, and peroxide precursors (e.g. percarbonate, perborate and peroxy-
30 carboxylic acids such as peracetic acid).

Applications

The proteases of the invention may typically be added as components of detergent compositions. The proteases may also be useful in removal of proteinaceous soiling.

35 Furthermore, the novel proteases described in this

invention may be used in the treatment of protein in process water containing hypochlorite, especially wherein the hypochlorite is present at a concentration of 1-10 ppm.

Detergent Compositions

5 According to the invention, the protease may be added as a component of a detergent composition. As such, it may be included in the detergent composition in the form of a detergent additive. The detergent composition as well as the detergent additive may additionally comprise one or more other
10 enzymes, such as lipases, amylases, cutinases, cellulases and oxidoreductases.

 In a specific aspect, the invention provides a detergent additive. The enzymes may be included in a detergent composition by adding separate additives containing one or more
15 enzymes, or by adding a combined additive comprising all of these enzymes. A detergent additive of the invention, i.e. a separated additive or a combined additive, can be formulated e.g. as granulates, liquids, slurries, etc. Preferred detergent additive formulations are granulates, in particular non-dusting
20 granulates, liquids, in particular stabilized liquids, slurries, or protected enzymes.

 Non-dusting granulates may be produced, e.g., as disclosed in US 4,106,991 and 4,661,452 (both to Novo Industri A/S) and may optionally be coated by methods known in the art.
25 Examples of waxy coating materials are poly(ethylene oxide) products (polyethyleneglycol, PEG) with mean molar weights of 1000 to 20000; ethoxylated nonylphenols having from 16 to 50 ethylene oxide units; ethoxylated fatty alcohols in which the alcohol contains from 12 to 20 carbon atoms and in which there
30 are 15 to 80 ethylene oxide units; fatty alcohols; fatty acids; and mono- and di- and triglycerides of fatty acids. Examples of film-forming coating materials suitable for application by fluid bed techniques are given in patent GB 1483591. Liquid enzyme preparations may, for instance, be stabilized by adding
35 a polyol such as propylene glycol, a sugar or sugar alcohol, lactic acid or boric acid according to established methods.

Other enzyme stabilizers are well known in the art. Protected enzymes may be prepared according to the method disclosed in EP 238,216.

The detergent composition of the invention may be in any convenient form, e.g. as powder, granules, paste or liquid. A liquid detergent may be aqueous, typically containing up to 70% water and 0-30% organic solvent, or non-aqueous.

The detergent composition comprises one or more surfactants, each of which may be anionic, nonionic, cationic, or zwitterionic. The detergent will usually contain 0-50% of anionic surfactant such as linear alkylbenzenesulfonate (LAS), alpha-olefinsulfonate (AOS), alkyl sulfate (fatty alcohol sulfate) (AS), alcohol ethoxysulfate (AEOS or AES), secondary alkanesulfonates (SAS), alpha-sulfo fatty acid methyl esters, alkyl- or alkenylsuccinic acid or soap. It may also contain 0-40% of nonionic surfactant such as alcohol ethoxylate (AEO or AE), carboxylated alcohol ethoxylates, nonylphenol ethoxylate, alkylpolyglycoside, alkyl dimethylamineoxide, ethoxylated fatty acid monoethanolamide, fatty acid monoethanolamide, or polyhydroxy alkyl fatty acid amide (e.g. as described in WO 92/06154).

The detergent composition may additionally comprise one or more other enzymes, such as amylases, lipases, cutinases, cellulases and oxidoreductases.

The detergent may contain 1-65% of a detergent builder or complexing agent such as zeolite, diphosphate, triphosphate, phosphonate, citrate, nitrilotriacetic acid (NTA), ethylenediaminetetraacetic acid (EDTA), diethylenetriaminepentaacetic acid (DTMPA), alkyl- or alkenylsuccinic acid, soluble silicates or layered silicates (e.g. SKS-6 from Hoechst). The detergent may also be unbuilt, i.e. essentially free of detergent builder.

The detergent may comprise one or more polymers. Examples are carboxymethylcellulose (CMC), poly(vinylpyrrolidone) (PVP), polyethyleneglycol (PEG), poly(vinyl alcohol) (PVA), polycarboxylates such as polyacrylates,

maleic/acrylic acid copolymers and lauryl methacrylate/acrylic acid copolymers.

The detergent may contain a bleaching system which may comprise a H_2O_2 source such as perborate or percarbonate which may be combined with a peracid-forming bleach activator such as tetraacetythylenediamine (TAED) or nonanoyloxybenzenesulfonate (NOBS). Alternatively, the bleaching system may comprise peroxyacids of e.g. the amide, imide, or sulfone type.

The enzymes of the detergent composition of the invention may be stabilized using conventional stabilizing agents, e.g. a polyol such as propylene glycol or glycerol, a sugar or sugar alcohol, lactic acid, boric acid, or a boric acid derivative as e.g. an aromatic borate ester, and the composition may be formulated as described in e.g. WO 92/19709 and WO 92/19708.

The detergent may also contain other conventional detergent ingredients such as e.g. fabric conditioners including clays, foam boosters, suds suppressors, anti-corrosion agents, soil-suspending agents, anti-soil redeposition agents, dyes, bactericides, optical brighteners, or perfume.

The pH (measured in aqueous solution at use concentration) will usually be neutral or alkaline, e.g. 7-11.

Particular forms of detergent compositions within the scope of the invention include:

- 1) A detergent composition formulated as a granulate having a bulk density of at least 600 g/l comprising

- linear alkylbenzenesulfonate (calculated as acid)	7 - 12%
- alcohol ethoxysulfate (e.g. C_{12-18} alcohol, 1-2 EO) or alkyl sulfate (e.g. C_{16-18})	1 - 4%
- alcohol ethoxylate (e.g. C_{14-15} alcohol, 7 EO)	5 - 9%
- sodium carbonate (as Na_2CO_3)	14 - 20%
- soluble silicate (as $Na_2O, 2SiO_2$)	2 - 6%
- zeolite (as $NaAlSiO_4$)	15 - 22%

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- sodium sulfate (as Na_2SO_4)	0 - 6%
- sodium citrate/citric acid (as $\text{C}_6\text{H}_5\text{Na}_3\text{O}_7/\text{C}_6\text{H}_8\text{O}_7$)	0 - 15%
- sodium perborate (as $\text{NaBO}_3 \cdot \text{H}_2\text{O}$)	11 - 18%
5 - TAED	2 - 6%
- carboxymethylcellulose	0 - 2%
- polymers (e.g. maleic/acrylic acid copolymer, PVP, PEG)	0 - 3%
- enzymes	0 - 5%
10 - minor ingredients (e.g. suds suppressors, perfume, optical brightener, photobleach)	0 - 5%

2) A detergent composition formulated as a granulate having
a bulk density of at least 600 g/l comprising

15 - linear alkylbenzenesulfonate (calculated as acid)	6 - 11%
- alcohol ethoxysulfate (e.g. C_{12-18} alcohol, 1-2 EO) or alkyl sulfate (e.g. C_{16-18})	1 - 3%
20 - alcohol ethoxylate (e.g. C_{14-15} alcohol, 7 EO)	5 - 9%
- sodium carbonate (as Na_2CO_3)	15 - 21%
- soluble silicate (as $\text{Na}_2\text{O} \cdot 2\text{SiO}_2$)	1 - 4%
- zeolite (as NaAlSiO_4)	24 - 34%
25 - sodium sulfate (as Na_2SO_4)	4 - 10%
- sodium citrate/citric acid (as $\text{C}_6\text{H}_5\text{Na}_3\text{O}_7/\text{C}_6\text{H}_8\text{O}_7$)	0 - 15 %
- carboxymethylcellulose	0 - 2%
- polymers (e.g. maleic/acrylic acid copolymer, 30 PVP, PEG)	1 - 6%
- enzymes	0 - 5%
- minor ingredients (e.g. suds suppressors, perfume)	0 - 5%

3) A detergent composition formulated as a granulate having

a bulk density of at least 600 g/l comprising		
- linear alkylbenzenesulfonate (calculated as acid)	5 - 9%	
5 - alcohol ethoxylate (e.g. C ₁₂₋₁₅ alcohol, 7 EO)	7 - 14%	
- soap as fatty acid (e.g. C ₁₆₋₂₂)	1 - 3%	
- sodium carbonate (as Na ₂ CO ₃)	10 - 17%	
- soluble silicate (as Na ₂ O, 2SiO ₂)	3 - 9%	
10 - zeolite (as NaAlSiO ₄)	23 - 33%	
- sodium sulfate (as Na ₂ SO ₄)	0 - 4%	
- sodium perborate (as NaBO ₃ ·H ₂ O)	8 - 16%	
- TAED	2 - 8%	
- phosphonate (e.g. EDTMPA)	0 - 1%	
15 - carboxymethylcellulose	0 - 2%	
- polymers (e.g. maleic/acrylic acid copolymer, PVP, PEG)	0 - 3%	
- enzymes	0 - 5%	
- minor ingredients (e.g. suds suppressors, 20 perfume, optical brightener)	0 - 5%	

4) A detergent composition formulated as a granulate having a bulk density of at least 600 g/l comprising

- linear alkylbenzenesulfonate (calculated as acid)	8 - 12%
25 - alcohol ethoxylate (e.g. C ₁₂₋₁₅ alcohol, 7 EO)	10 - 25%
- sodium carbonate (as Na ₂ CO ₃)	14 - 22%
- soluble silicate (as Na ₂ O, 2SiO ₂)	1 - 5%
- zeolite (as NaAlSiO ₄)	25 - 35%
30 - sodium sulfate (as Na ₂ SO ₄)	0 - 10%
- carboxymethylcellulose	0 - 2%

- polymers (e.g. maleic/acrylic acid copolymer,
PVP, PEG) 1 - 3%
- enzymes 0 - 5%
- minor ingredients (e.g. suds suppressors,
5 perfume) 0 - 5%

5) An aqueous liquid detergent composition comprising

- linear alkylbenzenesulfonate
(calculated as acid) 15 - 21%
- alcohol ethoxylate
10 (e.g. C₁₂₋₁₅ alcohol, 7 EO or
C₁₂₋₁₅ alcohol, 5 EO) 12 - 18%
- soap as fatty acid (e.g. oleic acid) 3 - 13%
- alkenylsuccinic acid (C₁₂₋₁₄) 0 - 13%
- aminoethanol 8 - 18%
- 15 - citric acid 2 - 8%
- phosphonate 0 - 3%
- polymers (e.g. PVP, PEG) 0 - 3%
- borate (as B₄O₇) 0 - 2%
- ethanol 0 - 3%
- 20 - propylene glycol 8 - 14%
- enzymes 0 - 5%
- minor ingredients
(e.g. dispersants, suds suppressors,
perfume, optical brightener) 0 - 5%

25 6) An aqueous structured liquid detergent composition comprising

- linear alkylbenzenesulfonate
(calculated as acid) 15 - 21%
- alcohol ethoxylate
30 (e.g. C₁₂₋₁₅ alcohol, 7 EO
or C₁₂₋₁₅ alcohol, 5 EO) 3 - 9%
- soap as fatty acid (e.g. oleic acid) 3 - 10%

- | | |
|--|----------|
| - zeolite (as NaAlSiO_4) | 14 - 22% |
| - potassium citrate | 9 - 18% |
| - borate (as B_4O_7) | 0 - 2% |
| - carboxymethylcellulose | 0 - 2% |
| 5 - polymers (e.g. PEG, PVP) | 0 - 3% |
| - anchoring polymers as
e.g. lauryl methacrylate/acrylic acid copolymer;
molar ratio 25:1; MW 3800 | 0 - 3% |
| - glycerol | 0 - 5% |
| 10 - enzymes | 0 - 5% |
| - minor ingredients
(e.g. dispersants, suds suppressors, perfume,
optical brighteners) | 0 - 5% |
| 7) A detergent composition formulated as a granulate having a | |
| 15 bulk density of at least 600 g/l comprising | |
| - fatty alcohol sulfate | 5 - 10% |
| - ethoxylated fatty acid monoethanolamide | 3 - 9% |
| - soap as fatty acid | 0 - 3% |
| - sodium carbonate (as Na_2CO_3) | 5 - 10% |
| 20 - soluble silicate (as $\text{Na}_2\text{O}, 2\text{SiO}_2$) | 1 - 4% |
| - zeolite (as NaAlSiO_4) | 20 - 40% |
| - sodium sulfate (as Na_2SO_4) | 2 - 8% |
| - sodium perborate (as $\text{NaBO}_3 \cdot \text{H}_2\text{O}$) | 12 - 18% |
| - TAED | 2 - 7% |
| 25 - polymers (e.g. maleic/acrylic acid copolymer,
PEG) | 1 - 5% |
| - enzymes | 0 - 5% |
| - minor ingredients (e.g. optical brightener,
suds suppressors, perfume) | 0 - 5% |
| 30 8) A detergent composition formulated as a granulate comprising | |
| - linear alkylbenzenesulfonate
(calculated as acid) | 8 - 14% |

- ethoxylated fatty acid monoethanolamide	5 - 11%
- soap as fatty acid	0 - 3%
- sodium carbonate (as Na_2CO_3)	4 - 10%
- soluble silicate (as $\text{Na}_2\text{O}, 2\text{SiO}_2$)	1 - 4%
5 - zeolite (as NaAlSiO_4)	30 - 50%
- sodium sulfate (as Na_2SO_4)	3 - 11%
- sodium citrate (as $\text{C}_6\text{H}_5\text{Na}_3\text{O}_7$)	5 - 12%
- polymers (e.g. PVP, maleic/acrylic acid copolymer, PEG)	1 - 5%
10 - enzymes	0 - 5%
- minor ingredients (e.g. suds suppressors, perfume)	0 - 5%

9) A detergent composition formulated as a granulate comprising

15 - linear alkylbenzenesulfonate (calculated as acid)	6 - 12%
- nonionic surfactant,	1 - 4%
- soap as fatty acid	2 - 6%
- sodium carbonate (as Na_2CO_3)	14 - 22%
- zeolite (as NaAlSiO_4)	18 - 32%
20 - sodium sulfate (as Na_2SO_4)	5 - 20%
- sodium citrate (as $\text{C}_6\text{H}_5\text{Na}_3\text{O}_7$)	3 - 8%
- sodium perborate (as $\text{NaBO}_3 \cdot \text{H}_2\text{O}$)	4 - 9%
- bleach activator (e.g. NOBS or TAED)	1 - 5%
- carboxymethylcellulose	0 - 2%
25 - polymers (e.g. polycarboxylate or PEG)	1 - 5%
- enzymes	0 - 5%
- minor ingredients (e.g. optical brightener, perfume)	0 - 5%

- 10) An aqueous liquid detergent composition comprising
- linear alkylbenzenesulfonate
(calculated as acid) 15 - 23%
 - alcohol ethoxysulfate
5 (e.g. C₁₂₋₁₅ alcohol, 2-3 EO) 8 - 15%
 - alcohol ethoxylate
(e.g. C₁₂₋₁₅ alcohol, 7 EO
or C₁₂₋₁₅ alcohol, 5 EO) 3 - 9%
 - soap as fatty acid (e.g. lauric acid) 0 - 3%
 - 10 - aminoethanol 1 - 5%
 - sodium citrate 5 - 10%
 - hydrotrope (e.g. sodium toluenesulfonate) 2 - 6%
 - borate (as B₄O₇) 0 - 2%
 - carboxymethylcellulose 0 - 1%
 - 15 - ethanol 1 - 3%
 - propylene glycol 2 - 5%
 - enzymes 0 - 5%
 - minor ingredients (e.g. polymers, dispersants,
perfume, optical brighteners) 0 - 5%
- 20 11) An aqueous liquid detergent composition comprising
- linear alkylbenzenesulfonate
(calculated as acid) 20 - 32%
 - alcohol ethoxylate
(e.g. C₁₂₋₁₅ alcohol, 7 EO
25 or C₁₂₋₁₅ alcohol, 5 EO) 6 - 12%
 - aminoethanol 2 - 6%
 - citric acid 8 - 14%
 - borate (as B₄O₇) 1 - 3%
 - polymer (e.g. maleic/acrylic acid copolymer,
30 anchoring polymers as e.g.
lauryl methacrylate/acrylic acid
copolymer and CMC) 0 - 3%
 - glycerol 3 - 8%

- enzymes 0 - 5%
- minor ingredients (e.g. hydrotropes, dispersants, perfume, optical brighteners) 0 - 5%

12) A detergent composition formulated as a granulate having
 5 a bulk density of at least 600 g/l comprising

- anionic surfactant (linear alkylbenzenesulfonate, alkyl sulfate, alpha-olefinsulfonate, alpha-sulfo fatty acid methyl esters, alkanesulfonates, soap) 25 - 40%
- 10 - nonionic surfactant (e.g. alcohol ethoxylate) 1 - 10%
- sodium carbonate (as Na_2CO_3) 8 - 25%
- soluble silicates (as Na_2O , 2SiO_2) 5 - 15%
- sodium sulfate (as Na_2SO_4) 0 - 5%
- 15 - zeolite (as NaAlSiO_4) 15 - 28%
- sodium perborate (as $\text{NaBO}_3 \cdot 4\text{H}_2\text{O}$) 0 - 20%
- bleach activator (TAED or NOBS) 0 - 5%
- enzymes 0 - 5%
- minor ingredients 0 - 3%
- 20 (e.g. perfume, optical brighteners)

13) Detergent formulations as described in 1) - 12) where the content of linear alkylbenzenesulfonate - or a part of it - is substituted by alkyl sulfate (C_{12} - C_{18}).

14) Detergent formulations as described in 1) - 13) which
 25 contain a stabilized or encapsulated peracid either as an additional component or as a substitute for already specified bleach systems.

15) Detergent compositions as described in 1), 3), 7), 9) and
 12) where the content of perborate is substituted with percar-
 30 bonate.

16) Detergent compositions as described in 1), 3), 7), 9) and 12) which additionally contains a Manganese catalyst. The Manganese catalyst may e.g. be one of the compounds described in "Efficient manganese catalysts for low-temperature bleaching", Nature 369, 1994, pp. 637-639.

17) Detergent composition formulated as a nonaqueous detergent liquid comprising a liquid nonionic surfactant as e.g. linear alkoxyated primary alcohol, a builder system (e.g. phosphate), enzyme and alkali. The detergent may also comprise anionic surfactant and/or a bleach system.

The present invention is further illustrated in the following examples which are not in any way intended to limit the scope of the invention as claimed.

EXAMPLE 1

Bacillus sp., DSM 8473, was cultivated at 37°C on a rotary shaking table (300 r.p.m.) in 500 ml baffled Erlenmeyer flasks containing 100 ml of medium of the following composition (per litre):

Potato starch	100	g
Ground barley	50	g
Soybean flour	20	g
Na ₂ HPO ₄ x 12 H ₂ O	9	g
Pluronic®	0.1	g
Sodium caseinate	10	g

The starch in the medium was liquified with α -amylase, and the medium was sterilized by heating at 120°C for 45 minutes.

After 3 days of incubation the proteolytic activity of the culture was determined using the method described above.

After cultivation, the enzyme activity of the broth was 120 CPU/l.

After separation of the solid material the protease was purified by a conventional chromatographic method and then

freeze-dried. The freeze-dried preparation had an activity of 7.4 CPU/g.

The characteristics of the preparation prepared in accordance with this Example have been referred to earlier in this specification, and reference is made hereto.

EXAMPLE 2

Stability Performance

The stability performance tests were conducted in 1.1 g/l of a commercial American powder detergent dissolved in approx. 6° dH (German Hardness) water with different concentrations of sodium hypochlorite at 25°C, isothermally for 60 minutes, with a protease concentration of 0.3 CPU per litre.

The results of these tests are shown in table 1 below:

Table 1

	0 ppm NaOCl	5 ppm NaOCl	10 ppm NaOCl
Alcalase	100%	10%	0%
Primase	100%	5%	0%
Esperase	55%	0%	0%
Savinase	100%	0%	0%
Durazyme	100%	40%	0%
NOVEL PROTEASE	100%	100%	30%

Table 1 shows that the novel protease has a much higher stability in solutions containing sodium hypochlorite than known proteases: The novel protease has a residual activity of 30% in a detergent solution containing 10 ppm NaOCl, whereas known proteases have absolutely no activity under these conditions.

EXAMPLE 3

The stability performance tests were also conducted in 1.1 g/l of a commercial American powder detergent dissolved in approx. 6° dH (German Hardness) water with 1% of Proxan (39.5% CH_3COOOH , 4.5% H_2O_2 , 44% CH_3COOH , 11.3% H_2O , 0.7% H_2SO_4) at 25°C, isothermally for 60 minutes, with a protease concentration of 0.3 CPU per litre.

The results of these tests are shown in table 2 below:

Table 2

10	Enzyme	Residual Activity
	Alcalase	25%
	Primase	20%
	Esperase	20%
	Savinase	25%
15	Durazyme	85%
	NOVEL PROTEASE	100%

Table 2 shows that the novel protease has a higher stability in solutions containing Proxan than known proteases.

EXAMPLE 4**20 Wash Performance**

The wash performance tests were accomplished on grass soiled cotton, at 20°C, isothermally for 10 minutes.

The tests were performed at enzyme concentrations of 0.0025, 0.005, 0.010, 0.050, 0.1, 0.2 and 0.5 CPU per litre.

25 2.0 g/l of a commercial American powder detergent were used. The detergent was dissolved in approx. 6° dH (German Hardness) water, and pH was adjusted to 9.5. The textile/wash liquor ratio was 6 g of textile per litre of wash liquor.

Subsequent to washing, the cloths were flushed in running
30 tap water and air-dried. The remission (%R) at 460 nm was determined.

As a measure of the wash performance differential remission, ΔR , was used being equal to the remission after wash with enzyme added, minus the remission after wash with no enzyme added.

5 The results of these tests are shown in Table 3 below
(mean of 2 tests):

Table 3

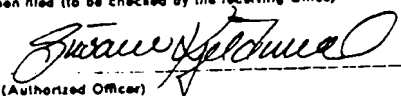
	NOVEL PROTEASE CONCENTRATION CPU/l	ΔR
10	0.0025	3.8
	0.005	6.6
	0.010	6.4
	0.050	11.8
	0.1	12.9
15	0.2	13.6
	0.5	13.1

Table 3 shows that the novel protease is well suited for use as a detergent enzyme.

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International Application No: PCT/

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MICROORGANISMS	
Optional Sheet in connection with the microorganism referred to on page <u>2</u> , line <u>7-8</u> of the description ¹	
A. IDENTIFICATION OF DEPOSIT ²	
Further deposits are identified on an additional sheet <input type="checkbox"/> ³	
Name of depositary institution ⁴	
DEUTSCHE SAMMLUNG VON MIKROORGANISMEN UND ZELL-KULTUREN GmbH	
Address of depositary institution (including postal code and country) ⁴	
Mascheroder Weg 1b, D-38124 Braunschweig, Federal Republic of Germany	
Date of deposit ⁵	Accession Number ⁶
23 August 1993	DSM 8473
B. ADDITIONAL INDICATIONS ⁷ (leave blank if not applicable). This information is continued on a separate attached sheet <input type="checkbox"/>	
In respect of those designations in which a European and/or Australian patent is sought, during the pendency of the patent application a sample of the deposited microorganism is only to be provided to an independent expert nominated by the person requesting the sample (Rule 28(4) EPC / Regulation 3.25 of Australia Statutory Rules 1991 No 71).	
C. DESIGNATED STATES FOR WHICH INDICATIONS ARE MADE ⁸ (If the indications are not for all designated States)	
D. SEPARATE FURNISHING OF INDICATIONS ⁹ (leave blank if not applicable)	
The indications listed below will be submitted to the International Bureau later ⁹ (Specify the general nature of the indications e.g., "Accession Number of Deposit")	
E. <input checked="" type="checkbox"/> This sheet was received with the international application when filed (to be checked by the receiving Office)	
 (Authorized Officer)	
<input type="checkbox"/> The date of receipt (from the applicant) by the International Bureau is ¹⁰	
was _____ (Authorized Officer)	

CLAIMS

1. A protease **characterized** by having immunochemical properties identical to those of a protease derived from the strain Bacillus sp., DSM 8473.
- 5 2. A protease according to claim 1, the protease retaining at least 80% of activity after 60 min. at 25°C in 5 ppm hypochlorite.
3. A protease according to claim 1, further characterized by:
 - 10 (a) An apparent molecular weight of approximately 30 kD as determined by SDS-PAGE;
 - (b) A pI of about 8.8 as determined by isoelectric focusing on LKB Ampholine PAG plates;
 - (c) Activity optimum at temperatures in the range from
15 50°C to 60°C, around 60°C, determined at pH 9.5 with casein as substrate; and
 - (d) More than 80% of activity in the range pH 6-11 when measured at 25°C with casein as substrate.
4. A biologically pure culture of a strain of Bacillus
20 sp., characterized by having the ability to produce a protease according to claim 1.
5. A culture according to claim 4, the strain being Bacillus sp., DSM 8473, or a mutant or a variant thereof.
6. A process for the preparation of a protease according
25 to any of claims 1-3, which process comprises cultivation of a protease producing strain of Bacillus sp. according to either of claims 4-5, in a suitable nutrient medium, containing carbon and nitrogen sources and inorganic salts, followed by recovery of the desired enzyme.
- 30 7. A process according to claim 6, wherein the strain is

Bacillus sp., DSM 8473, or a mutant or a variant thereof.

8. Use of a protease according to any of claims 1-3 in the treatment of protein in process water containing hypochlorite.

9. Use of a protease according to claim 8, wherein
hypochlorite is present at a concentration of 1-10 ppm.

10. Use of a protease according to any of claims 1-3 in removal of proteinaceous soiling.

11. A detergent composition comprising a protease according to any of claims 1-3 and a surfactant.

10 12. A detergent composition according to claim 11, which further comprises one or more other enzymes, in particular an amylase, a lipase, a cutinase, a cellulase and/or an oxidoreductase.

13. A detergent additive comprising a protease according
15 to any of claims 1-3, provided in the form of a non-dusting granulate, a stabilized liquid, a slurry, or a protected enzyme.

14. A washing process comprising treatment of soiled fabric with protease according to any of claims 1-3.

20 15. A washing process according to claim 14, comprising a detergent additive according to claim 13.

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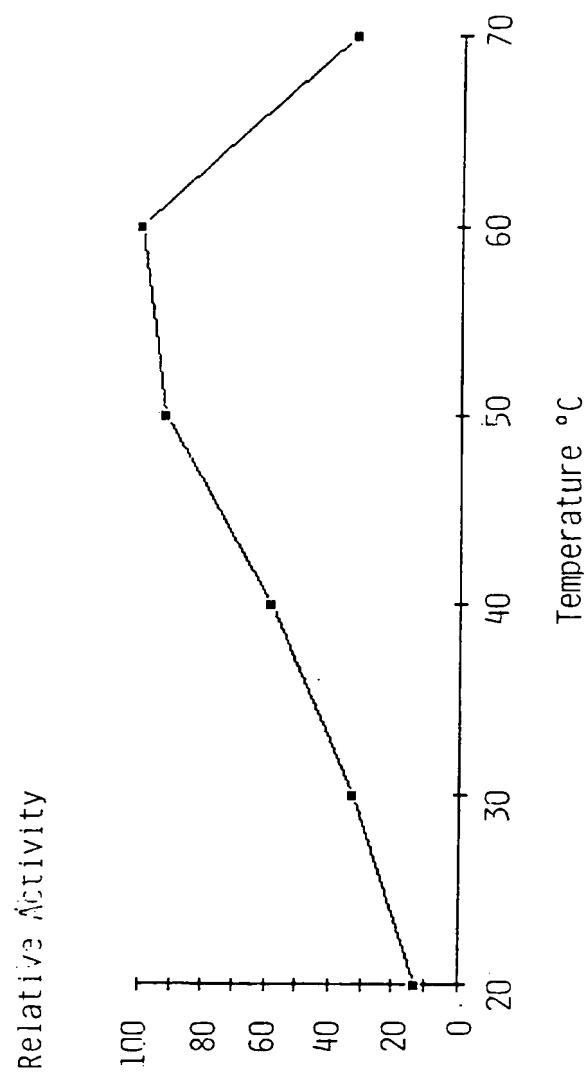


Fig. 1

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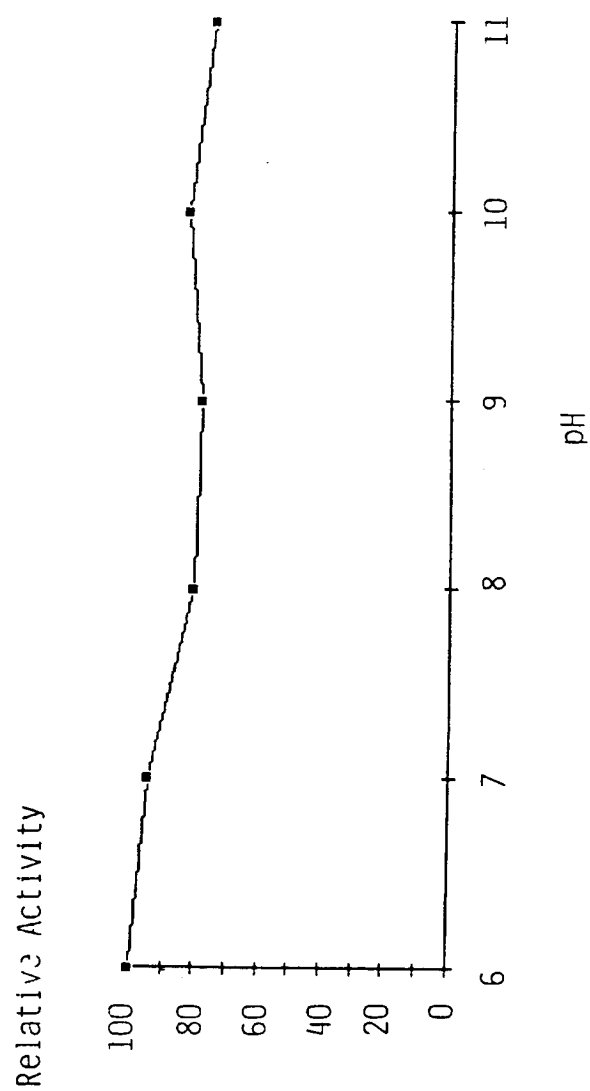


Fig. 2

INTERNATIONAL SEARCH REPORT

International application No.

PCT/DK 94/00331

A. CLASSIFICATION OF SUBJECT MATTER		
IPC6: C12N 9/54, C11D 3/386 According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED		
Minimum documentation searched (classification system followed by classification symbols)		
IPC6: C12N		
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched		
SE,DK,FI,NO classes as above		
Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)		
CA, IFIPAT, WPIDS		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	WO, A1, 8801293 (NOVO INDUSTRI A/S), 25 February 1988 (25.02.88) --	1-7, 11-15
X	WO, A1, 9207067 (NOVO NORDISK A/S), 30 April 1992 (30.04.92) --	1, 3-7, 11-15
A	US, A, 5118623 (GEORGE BOGUSLAWSKI ET AL), 2 June 1992 (02.06.92) -- -----	1-15
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input checked="" type="checkbox"/> See patent family annex.		
* Special categories of cited documents: "A" document defining the general state of the art which is not considered to be of particular relevance "E" earlier document but published on or after the international filing date "L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified) "O" document referring to an oral disclosure, use, exhibition or other means "P" document published prior to the international filing date but later than the priority date claimed "T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention "X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone "Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art "&" document member of the same patent family		
Date of the actual completion of the international search		Date of mailing of the international search report
15 December 1994		19 -12- 1994
Name and mailing address of the ISA/ Swedish Patent Office Box 5055, S-102 42 STOCKHOLM Facsimile No. +46 8 666 02 86		Authorized officer Carolina Gomez Lagerlöf Telephone No. +46 8 782 25 00

INTERNATIONAL SEARCH REPORT
Information on patent family members

26/11/94

International application No.
PCT/DK 94/00331

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO-A1- 8801293	25/02/88	DE-A, T- 3787009 EP-A- 0277216 JP-T- 1500642	16/09/93 10/08/88 09/03/89
WO-A1- 9207067	30/04/92	EP-A- 0552222 JP-T- 6503717 US-A- 5358865	28/07/93 28/04/94 25/10/94
US-A- 5118623	02/06/92	NONE	

